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### Go associates with another 40 kDa brain protein

#### Bradley M. Denker and Eva J. Neer

Harvard Medical School and Department of Medicine, Brigham and Women's Hospital, Boston, MA, USA

#### Received 6 November 1990

Guanine nucleotide binding proteins (G proteins) mediate a variety of cellular responses to external stimuli. Pure G protein, receptor, and effector are sufficient to reconstitute hormonal activation of an effector in phospholipid vesicles, but other components may be important for specificity or localization in vivo. If another protein associates with G<sub>s</sub>, the molecular weight of G<sub>s</sub> solubilized from membranes would be larger than the molecular weight of G<sub>s</sub> after purification. We find that G<sub>s</sub> solubilized from bovine brain membranes by Triton X-100 behaves as a single population of molecules on sucrose density gradients and gel filtration columns. Its molecular mass is about 40 kDa larger than pure G<sub>s</sub>. Association of G<sub>s</sub> with the other protein is fragile as the proteins dissociate on further purification. There was no difference in ADP-ribosylation or tryptic cleavage of G<sub>s</sub> in larger and smaller form. These studies provide a basis for future experiments to stabilize the interaction and identify the protein.

O protein; Signal transduction; Sucrose gradient; Stokes radii; Molecular weight

#### I. INTRODUCTION

A variety of cellular responses to external stimuli are mediated by a group of guanine nucleotide binding proteins (G proteins) made up of  $\alpha$ ,  $\beta$ , and  $\gamma$  subunits. The basic components of this signal transduction system are a receptor, an effector, and the G protein that couples the two. Incorporation of pure receptor, G protein and effector into lipid vesicles is sufficient for agonist stimulation of the effector. For example, isoproterenol can stimulate adenylyl cyclase when pure  $\beta$ -adrenergic receptor, G<sub>s</sub>, and adenylyl cyclase are incorporated into phospholipid vesicles [1]. While such reconstitution studies reveal the minimum number of components. they cannot reveal whether other proteins interact with the signalling system. Such interactions may modulate the strength or specificity of responses, or localize the signal transducing molecule to membrane domains. For example, in some neural cells,  $\alpha_0$  is localized to growth cones where it interacts with GAP 43, an intracellular protein that can modulate GTP<sub>2</sub>S binding to G<sub>0</sub> [2]. Similarly, calmodulin interacts with G proteins and the catalytic unit of adenylyl cyclase [3,4]. Cytoskeletal proteins can also interact with G proteins and other parts of the transduction system [5-11].

Although purification of the components is essential for precise analysis of their structure and function, there is a danger that during purification loosely bound regulatory or localizing proteins may be lost. As an initial step in determining whether other proteins

Correspondence address: B.M. Denker, Renal Division, Brigham and Women's Hospital, 75 Francis Street, Boston, MA 02115, USA

Abbreviations: GDP\$s, guanosine 5'-O-(2-thiodiphosphate)

associate with G proteins, we asked whether the molecular weight of the Go as first solubilized from the membrane is larger than the molecular weight of the pure protein. A larger molecular weight would suggest that Go is associating with another protein. We measured the molecular weight of Go in unfractionated solubilized membranes by the hydrodynamic techniques of sucrose density centrifugation and gel filtration. As long as a particular protein complex in a mixture of proteins behaves as a single hydrodynamic particle, its molecular weight can be explicitly determined. To identify Go, we analyzed fractions from sucrose density gradients and gel filtration by Western blots probed with antibody to  $\alpha_0$  and  $\beta$ . These studies show that the whole population of Go molecules is initially solubilized from brain membranes as a complex with another protein.

#### 2. MATERIALS AND METHODS

quent analysis.

Calf brains (obtained from a local slaughter house) were dissected on ice, and the cortices (80 g) were homogenized in 400 ml 50 mM Tris (pH 8.0), 75 mM sucrose, 1 mM EDTA, 6 mM MgCl<sub>2</sub>, 1  $\mu$ g/ml soybean and lima bean trypsin inhibitors, 3 mM benzamidine, 1 mM dithiothreitol (buffer A). All reagents in buffer A were from Sigma. The homogenate was centrifuged at  $10\,000\,\times g$  for 20 min at  $4^{\circ}$ C, the pellet resuspended in 350 ml buffer A and recentrifuged at  $10\,000\,\times g$  for 20 min at  $4^{\circ}$ C. The pellet was resuspended in 100 ml of buffer A and frozen at  $-70^{\circ}$ C. Crude membranes (1 ml) were mixed with 1 ml buffer A containing 2  $\mu$ M GDP $\beta$ S and 2% Triton X-100 (Fisher) and gently shaken at  $4^{\circ}$ C for 90 min, then centrifuged at  $22\,000\,\times g$  for 70 min at  $4^{\circ}$ C. The supernatant (5.6 mg/ml) was collected for subse-

2.1. Triton X-100 solubilized bovine brain membrane preparation

2.2. Physical properties of Triton-solubilized G<sub>0</sub> and purified G<sub>0</sub>. The sedimentation coefficient and v were determined as previously described (12]. The 5-20% sucrose gradients were made up in buffer A with 0.3% Triton X-100 and were internally calibrated with BSA s<sub>20,w</sub>, 4.31 S. v, 0.732 cm<sup>3</sup>/g, carbonic anhydrase, s<sub>20,w</sub>, 2.85 S, v,

0.735 cm<sup>2</sup>/g, and catalase  $x_{16,n}$  11.3 S,  $\theta$ , 0.730 cm<sup>2</sup>/g. The Stokes radius was determined by get filtration on Ultroget ACA-34 (LKB) columns equilibrated with buffer A, 0.3% Triton X-100 and internally calibrated with £-galactosidase (69 Å), BSA (37 Å), ovalbumin (27 Å), and catalase (52 Å).

#### 2.3. Mixeellaneous methods

Purified G., was prepared as previously described [13]. Protein was determined by the method of Lowry [14] and modified by Bailey [15]. SDS-PAGE was carried out as described by Laemmli [16], and Western blots performed by the method of Towbin [17] as described previously [12]. The anti-m, and anti-f antibodies have been previously described from this laboratory [12]. ADP-ribosylation by pertussis toxin (List Biological) was carried out as previously described [12] for 10-30 min in the presence of 0-10 µM GTP<sub>2</sub>S. Tryptic digestion was carried out as previously described [18] for 10-30 min in the presence of 0-10 µM GTP<sub>2</sub>S.

#### 3. RESULTS

### 3.1. G<sub>0</sub> is monodisperse in Triton X-100-solubilized brain membranes

Fig. 1 shows that all of the Triton-soluble  $G_0$  applied to a sucrose density gradient sediments as a single peak. On sucrose gradients the peak width at half-height for Triton-soluble  $G_0$  is the same as for the marker proteins. The ratio of peak width for Triton-soluble  $G_0$  to the peak width for marker proteins was  $1.05 \pm 0.24$  (n=13). By gel filtration, Triton-soluble  $G_0$  also clutes as a single peak. The ratio of peak width for Triton-soluble  $G_0$  to peak width for marker proteins was  $0.85 \pm 0.19$  (n=6). The observation that there is only one peak of  $G_0$  that is no wider than a peak of pure protein suggests that  $G_0$  makes up a single population of molecules and is not randomly aggregating with other proteins.

# 3.2. The molecular weight of Triton X-100-solubilized $G_0$ is larger than purified $G_0$

To compare the molecular weight of unfractionated solubilized  $G_0$  to purified  $G_0$ , we measured the sedimentation coefficient  $(s_{20,w})$ , Stokes radius, and partial specific volume  $(\bar{v})$  of both preparations in parallel. The

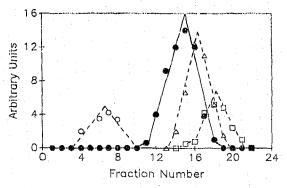


Fig. 1. Elution profile of Triton-solubilized G<sub>2</sub> on 5-20% sucrose gradient. Bovine brain membranes were prepared and analyzed ās described in Materials and Methods. Coomassic-stained gels and Western blots were scanned with an LKB Laser Densitometer and arbitrary units plotted versus fraction number. (Ο) catalase; (Δ) BSA; (□) carbonic anhydrase; and (③) G<sub>0</sub>.

Table 1

Physical properties of Trium X-100 solubilized and purified G<sub>a</sub> in the presence of 1 aM GDP3S

Parameter*	Triton X-100 vol.				Purified C.			
Partial specific volume, v, (ml/y)*	0,80	×	0,03	(13)	0.77	A	0.02	(13)
Sedimentation coeffi- cient, x <sub>im,w</sub> (S)	5.6	Æ	6,0	(H)	4.5	*	6,0	(11)
Stokes radius, a (A)	67	*	3	(6)	44	*	4	(6)
Molecular weight	212 000				124 000			
M.	(19) 000=233 00014				(105 000-139 000)4			
Triton X-100 bound, (mu/mu protein)*	0.32				0.17			
Molecular weight of	144 000				1010	ÓÖ		
protein	(130000=158000)				( 87 000-115 000)			

A The values are calculated from all possible paired combinations of data from H<sub>2</sub>O and D<sub>2</sub>O gradients, and the results given are the averages ± SE. The 1 umber in parenthesis is the sum of experiments in H<sub>2</sub>O and D<sub>2</sub>O

The values of Pwere calculated from H<sub>2</sub>O and D<sub>2</sub>O gradients by the method of Sadler [24]

\* Molecular weight was calculated according to the following equa-

$$M_1 = \frac{6 \pi N \eta_{20,w}}{1 - (9 Q_{20,w})} a_{525,w}$$

where N is Avogadro's number,  $n_{20,w}$  is the viscosity of water at  $20^{\circ}$ C,  $\rho_{20,w}$  is the density of water at  $20^{\circ}$ C

Range values were calculated using 1 SE above and below each parameter

\* Calculated using Pprotein = 0.735 and PTition = 0.94 [25]

partial specific volume was determined by the position of Go relative to markers in H2O and D2O sucrose density gradients. Table I shows the measured parameters and calculated molecular weight of Triton-solubilized Go and purified Go. There was no significant difference in the partial specific volume for Triton X-100 solubilized Go and purified Go. Both s20,w and Stokes radius were larger for Triton-solubilized Go than for purified Go. The values obtained for pure Go are in agreement with those previously reported [12]. The molecular weight of Triton-solubilized Go (detergent plus protein) is 212000. However, 32% of this mass is detergent so that the molecular weight of the protein is 144000. The molecular weight of pure Go detergent protein complex is 124000 with 17% Triton bound giving a molecular weight of 101000 for the protein portion. The molecular weight predicted from the sum of the components for  $\alpha, \beta$ , and  $\gamma$  is 85000. The difference in molecular weight between Triton-soluble Go and pure Go is 43 000 and represents the total mass of material associating with G<sub>0</sub> in Triton-solubilized membranes.

The additional mass most likely represents a protein component\*. If it were carbohydrate only, the  $\bar{\nu}$  would be smaller than that of pure  $G_0$  while this mass of lipid would greatly increase the  $\bar{\nu}$ . The larger form of  $G_0$  is

<sup>\*</sup>GAP-43, a growth cone protein, is not the associating protein based on Western blots of sucrose density gradients of G<sub>0</sub> and GAP-43

unlikely to be  $\alpha \mathcal{S}_{\gamma}$  or  $\alpha(\mathcal{S}_{\gamma})_2$  because the ratio of immunoreactivity of  $\alpha$  to  $\beta$  was the same for Triton-soluble  $G_0$ , with a larger molecular weight and partially purified smaller  $G_0$ .

We next wanted to determine whether the larger form of  $G_a$  could be maintained through further steps of purification. One passage over an ion exchange column or repeat sucrose density centrifugation caused dissociation of the complex. The larger Stokes radius was preserved on repeat gel filtration but there was no further purification. These observations show that the complex of  $G_b$  and is associated protein is a fragile one which survives few steps of purification.

# 3.3. Functional comparison of larger and smaller forms of G<sub>o</sub>

To determine whether there are differences in the function of the larger and smaller forms of  $G_0$  we examined two functions unique to G proteins: the pertussis toxin-catalyzed inhibition by  $GTP_{\gamma}S$  of ADPribosylation [19] and  $GTP_{\gamma}S$ -dependent changes in the pattern of cleavage by trypsin [20]. There was no difference between the larger and smaller forms of  $G_0$  in the  $GTP_{\gamma}S$  dose/response curve for either of these processes. G protein functions such as GTP binding or GTP as activity cannot be compared because of other enzymes in the crude mixture.

#### 4. DISCUSSION

The protein which associates with  $G_o$  must be present in brain in an amount at least equal to that of  $G_o$  itself. We base this conclusion on the observation that  $G_o$  behaves as a single population of molecules with a molecular weight approximately 40 kDa larger than the sum of  $\alpha\beta\gamma$ . Since  $G_o$  makes up about 1% of brain membrane protein [12] the protein that associates with it must also be extremely abundant. Its association with  $G_o$  is fragile and the two proteins are separated early in the purification of  $G_o$ . Nevertheless, the presumed abundance of the protein makes brain a reasonable source for its eventual purification.

Early studies of the structure of unpurified  $G_s^{**}$  also gave values for the molecular weight which were 30-100 kDa larger than the value we can now calculate for  $\alpha_s \beta \gamma$  [21-23].  $G_s$  is a quantitatively minor membrane protein so any proteins associated with it might also be rare and consequently hard to identify. Our studies now show that even when the G protein is extremely abundant, its initial molecular weight is larger than that of the pure protein. We predict that other G proteins are also solubilized as specific complexes with another protein.

There is a growing body of evidence that signal transduction systems are more complicated than

\*\*G, is the stimulatory G protein of adenylyl cyclase and some calcium channels

originally proposed. The results reported here indicate that there is an association between G<sub>n</sub> and another brain protein with a total molecular weight of approximately 43 kDa. This protein may, of course, itself be made up of several subunits. The role this protein may be playing in modulating the signal transduction system or localizing it to membrane domains has yet to be elucidated. The present studies provide a basis for future experiments to stabilize the interaction and identify the protein.

Acknowledgements: The work described was supported by NIH Grant GM36259 to E.J.N. B.D. was supported by a NRSA Fellowship (GM12578).

#### REFERENCES

- May, D.C., Ross, E.M., Gilman, A.G. and Smigel, M.D. (1985)
   J. Biol, Chem. 260, 15829-15833.
- [2] Strittmatter, S.M., Valenzuela, D., Kennedy, T.E., Neer, E.J. and Fishman, M.C. (1990) Nature 344, 836-841.
- [3] Asano, T., Ogasawara, N., Kitajima, S. and Sano, M. (1986) FEBS Lett. 203, 135-138.
- [4] Salter, R.S., Krinks, M.H., Klee, C.B. and Neer, E.J. (1981) J. Biol. Chem. 256, 9830-9833.
- [5] Schlegel, W., Kempner, E.S. and Rodbell, M. (1979) J. Biol. Chem. 254, 5168-5176.
- [6] Dufau, M.L., Horner, K.A., Hayashi, K., Tsuruhara, T., Conn, P.M. and Catt, K.J. (1978) J. Biol. Chem. 253, 3721-3729.
- [7] Sahyoun, N.E., LeVine, III, H., David, J., Hebdon, G.M. and Cuatrecasas, P. (1981) Proc. Natl. Acad. Sci. USA 78, 6158-6162.
- [8] Cherskey, M.B.C., Zadunaisky, J.A. and Murphy, R.B. (1980) Proc. Natl. Acad. Sci. USA 77, 6401-6405.
- [9] Insel, P.A. and Kennedy, M.S. (1978) Nature 273, 471-473.
- [10] Rasenick, M.M., Stein, P.J. and Bittensky, M.L.J. (1981) Nature 294, 560-562.
- [11] Carlson, K.E., Woolkalis, M.J., Newhous, M.G. and Manning, D.R. (1986) Mol. Pharmacol. 30, 463-468.
- [12] Huff, R.M., Axton, J.M. and Neer, E.J. (1985) J. Biol. Chem. 260, 10864-10871.
- [13] Neer, E.J., Lok, J.M. and Wolf, L.G. (1984) J. Biol. Chem. 259, 14222-14229.
- [14] Lowry, O.H., Rosebrough, N.J., Farr, A.L. and Randall, R.J. (1951) J. Biol. Chem. 193, 265-275.
- [15] Bailey, J.L. (1967) in: Techniques in Protein Chemistry, p. 340, Elsevier, New York.
- [16] Laemmli, U.K. (1980) Nature 227, 680-685.
- [17] Towbin, H., Staehelin, T. and Gordon, J. (1979) Proc. Natl. Acad. Sci. USA 76, 4350-4354.
- [18] Neer, E.J., Pulsifer, L. and Wolf, L.O. (1988) J. Biol. Chem. 263, 8996-9000.
- [19] Tsai, S., Adamik, R., Kanaho, Y., Hewlett, E.L. and Moss, J. (1984) J. Biol. Chem. 259, 15320-15323.
- [20] Winslow, J. W., Van Amsterdam, J.R. and Neer, E.J. (1986) ... Biol. Chem. 261, 7571-7579.
- [21] Neer, E.J. and Salter, R.S. (1981) J. Biol. Chem. 256, 12102-12107.
- [22] Howlett, A.C. and Gilman, A.G. (1980) J. Biol. Chem. 255, 2861-2866.
- [23] Kaslow, H.R., Johnson, G.L., Brothers, V.M. and Bourne, H.R. (1980) J. Biol. Chem. 255, 3736-3741.
- [24] Sadler, J.E. (1979) J. Biol. Chem. 254, 4443.
- [25] Steele, J.C.H., Tanford, C. and Reynolds, J.A. (1978) Methods Enzymol. 48, 11-23.